# **BOOK OF ABSTRACTS**

FOA9

9th international conference on

**Fundamentals of Adsorption** 

May 20-25, 2007 Giardini Naxos, Sicily - Italy

#### SILICA WITH IMMOBILIZED PROPYLTHIOETHYLEAMINE AS AN ADSORBENT 251 FOR SELECTIVE DETERMINATION OF GOLD, PALLADIUM AND SILVER

O.P. Konoplitska, V. Zaitsev and G.M. Zaitseva

STRUCTURAL AND ADSORPTION CHARACTERISTICS OF TITANATE 252 NANOTUBE

Z.M. Wang, M. Wei, H.S. Zhou, H. Takagi and H. Hatori

SYNTHESIS OF TITANIA-BRIDGED GRAPHENE-LAYERED MATERIALS 253

Y.J. Liu, M. Aizawa, Z.M. Wang, H. Hatori and T. Hirotsu

SYNTHESIS AND CHARACTERIZATION OF CARBON NANOTUBE MEMBRANES 254

E. Vermisoglou, G. Pilatos, G. Romanos, N. Boukos and N. Kanellopoulos

OPTIMIZATION OF SYNTHESIS PROCEDURES FOR STRUCTURED PSA 255 ADSORBENTS

÷.,

A. Mosca, J. Hedlund, F. Ridha and P. Webley

ADSORPTION TO DETERMINING THE FRACTION OF OPEN CARBON 256 NANOTUBES IN SAMPLES

S. Agnihotri and J. Mota

EQUILIBRIA FOR ADSORPTION OF VALUABLE METALS ON CROSS-LINKED 257 CHITOSAN FIBER AND HIGHLY PORUS CHITOSAN BEAD

H. Nakamura, Miyasaka and H. Yoshida

STUDY ON ADSORPTION EQUILIBRIUM OF H2/N2/CH4/CO2 MIXTURE ON 258 ACTIVATED CARBON

J. Wu, L. Zhou, W. Su, Y. Sun and Y.P. Zhou

FLOW-THROUGH PORE STRUCTURE CHARACTERIZATION OF SILICA 259 MONOLITHS BY MERCURY INTRUSION AND RETRACTION; TRANSMISSION ELECTRON MICROSCOPY; HYDRODYNAMIC CHROMATOGRAPHY AND HYDRODYNAMIC PERMEABILITY

R. Skudas, B.A. Grimes, D. Lubda and K.K. Unger

#### Liquid phase adsorption

REMOVAL OF METHYLENE BLUE DYE FROM ITS AQUEOUS SOLUTION BY FRUIT SHELL OF AEGLE MERMELOSE AS ALTERNATIVE LOW COST ADSORBENT

T.K. Sen, A.A.B. Omar, S. Mishra, R. Patel and R. Dev Jain

ADSORPTION OF PLASMID DNA ON POROUS PARTICELS AND MONOLITHS	261

T. Tarmann and A. Jungbauer

BLANK CORRECTION IN DYNAMIC COLUMN BREAKTHROUGH EXPERIMENTS 262
A. Rajendran and S. Farooq

THE ROLE OF WATER AND SURFACE ACIDITY ON THE REACTIVE 263 ADSORPTION OF AMMONIA ON MODIFIED ACTIVATED CARBONS

### T. Bandosz and L.M. Le Leuch

REMOVAL OF ARSENIC (V) ONTO CHITOSAN: FROM SORPTION MECHANISM 264 EXPLANATION TO DYNAMIC WATER TREATMENT PROCESS

C. Gerente, Y. Andres, G. Mc Kay and P. Le Cloirec

ADSORPTION OF A BASIC DYE FROM AQUEOUS SOLUTION ONTO A NATURAL 265 ZEOLITE FIXED BED: SIMULATION AND OPERATION

B. Zarenezhad

SINGLE AND BINARY ADSORPTIONS OF HEAVY METALS ON PALM KERNEL 266 SHELL ACTIVATED CARBON

T. Choong, M. Muhammad, T.G. Chuah, R. Yunus, S. Abdul Rashid and D. Darmadi

BIOSORPTION OF CD(II), CR(VI), PB(II) ON AZADIRACHTA INDICA LEAF 267 POWDER

A. Sarma and K.G. Bhattacharyya

## SILICA WITH IMMOBILIZED PROPYLTHIOETHYLEAMINE AS AN ADSORBENT FOR SELECTIVE DETERMINATION OF GOLD, PALLADIUM AND SILVER

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Keywords: organosilica; immobilised complexes, trace analysis

Noble metals, particularly palladium, gold and silver, find an extensive use in the electrical industry, medical and jeweller industry. All these metals are very important elements for utilization and particularly from secondary sources since Ukraine has no deposits. One form possible sources is wastewater from non-ferrous metals industry.

The direct application of flame atomic absorption spectrometry to the detection of noble metals in complex matrices such as wastewater, geological and environmental samples is limited by numerous interferences, both mutual and those from associated base metals. For instance, base elements, i.e. zinc, copper, nickel and other, interfere on the signals atomic absorption determination of precious metals. Therefore, low concentrations of noble metals a separation/pre-concentration procedure prior to detection is required. For this purpose solvent extraction and solid phase extraction techniques (SPE) have been widely used. SPE has found increasing application for preconcentration of noble metals.

In the present study, silica with covalently immobilized propylthioethyleamine  $(SN-SiO_2)$  was used as SPE for selective pre-concentration of trace amount of palladium from wastewater, silver from solution of electrical contact and gold in acid-treated integral microcircuit. A pre-concentration method is based on adsorption of Pd<sup>2+</sup>, Au<sup>3+</sup> and Ag<sup>+</sup> on modified silica due to complex formation with immobilized organic ligand, followed by eluted and further determination by atomic-absorption technique.

In this study, various parameters, i.e. pH, time of stirring, volume of sample, chooses eluents and adsorption isotherm of silica gel has been studied. The influence of interfering elements in analyzed solutions on Pd, Au and Ag determination was investigated by using the development method. The results showed that up to 100  $\mu$ g ml<sup>-1</sup>Pb<sup>2+</sup>, 1000 $\mu$ g ml<sup>-1</sup>Cu<sup>2+</sup> did not interfere with the palladium, silver and gold signal. The results obtained are demonstrated that SN-SiO<sub>2</sub> can be used for selective pre-concentration of mentioned noble metals from very complex solutions. The method has been successfully applied to the analysis with detection limit 0.02  $\mu$ g ml<sup>-1</sup>, 0.01  $\mu$ g ml<sup>-1</sup> and 0.05  $\mu$ g ml<sup>-1</sup> of Ag, Au and Pd respectively.